

# The First Epoxidations of 1-Amidoallenes. A General Entry to Nitrogen-Substituted Oxyallyl Cations in Highly Stereoselective [4 + 3] Cycloadditions<sup>†</sup>

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Allenes are important synthons in organic synthesis.<sup>1</sup> Although epoxidations of allenes<sup>1,2</sup> and allenol ethers<sup>3,4</sup> have been reported, epoxidation of nitrogen-substituted allenes remains unexplored.<sup>1</sup> We envisaged that epoxidation of allenamides **1** could lead to the chiral allene oxides **2**, thereby providing a completely novel and general entry to chiral nitrogen-substituted [or stabilized] oxyallyl cations **3** for stereoselective [4 + 3] cycloadditions [Scheme 1].<sup>2,5</sup> The versatility of oxyallyl cations, especially of those heteroatom substituted, in highly regio- and stereoselective [4 + 3] cycloaddition reactions, leading to useful carbo- and heterocyclic systems, has attracted much attention from the synthetic community.<sup>6,7</sup> While significant advances have been made using oxygen,<sup>8</sup> sulfur,<sup>9</sup> or halogen-substituted<sup>10</sup> oxyallyl cations, nitrogen-substituted oxyallyls have received less attention.<sup>7,11–13</sup> The trivalency of the nitrogen atom renders nitrogen-substituted oxyallyl cations very attractive for developing stereoselective protocols [see the **R\*** group on nitrogen], which remains an important challenge in advancing the oxyallyl [4 + 3] cycloaddition.<sup>14</sup> Our interests in chiral allenamides **1**<sup>15,16</sup>

<sup>†</sup> With deepest respect and appreciation this paper is dedicated to Professor Gilbert Stork on the occasion of his 80th birthday.

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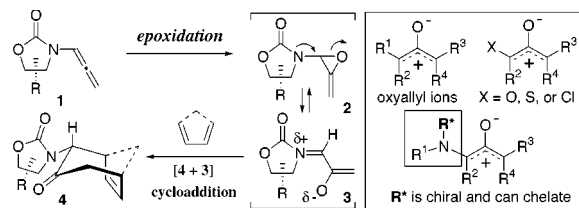
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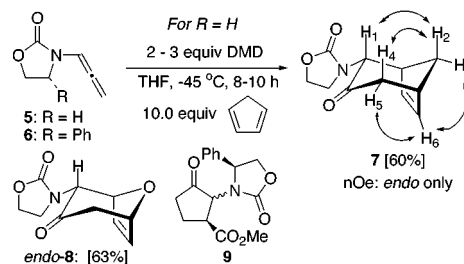
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## Scheme 1



## Scheme 2



allowed us to link together the epoxidation chemistry of allenamides with stereoselective [4 + 3] cycloadditions. We report here the first epoxidations of 1-amidoallenes and their applications as nitrogen-substituted oxyallyls in highly stereoselective [4 + 3] cycloadditions.

Dimethyldioxirane [DMD] was found to be the most useful protocol in epoxidizing allenamide **5** at low temperatures. The epoxidized intermediate could be readily trapped in the presence of 10.0 equiv of cyclopentadiene leading to the cycloadduct **7** in 60% yield as a single diastereomer that was assigned as *endo* [or compact<sup>6</sup>] via nOe experiments [Scheme 2].<sup>17</sup> DMD oxidation was also found to be very selective for the allenamide **5** in the presence of 10.0 equiv of furan, and subsequent cycloaddition led to isolation of the cycloadduct **8**<sup>18</sup> in 63% yield. This protocol establishes an attractive one-pot sequential process of epoxidation and cycloaddition. Reaction of **6** with methyl acrylate did not yield the [3 + 2] cycloadduct **9**, thereby suggesting that the nitrogen-substituted allene oxide intermediate **2** and/or oxyallyl cation **3** are electrophilic in character as expected.<sup>2</sup>

[4 + 3] cycloaddition reaction via epoxidation of chiral allenamide **6** was examined in detail. As shown in Table 1, this sequential epoxidation–[4 + 3] cycloaddition protocol could proceed in a range of different solvents<sup>19</sup> [entries 1–4], leading to the cycloadduct **10** in good yields. Although diastereoselectivity was modest, only *endo* isomers were observed, and stereochemistry of the major isomer **10a** [*endo-1*] was confirmed by X-ray structure. While there is a slight temperature effect on the stereochemical outcome [entries 5 and 6], the best diastereoselectivity was obtained when reactions were carried out in the presence of 2.0 equiv of ZnCl<sub>2</sub> [entries 9–11]. Notably at –78 °C [entry 11], the cycloadduct **10a** was isolated as a single diastereomer. Additives such as LiClO<sub>4</sub> and MgBr<sub>2</sub> [entries 7 and 8] were not useful, although they have been utilized in other oxyallyl [4 + 3] cycloadditions.<sup>8,10,12,14</sup>

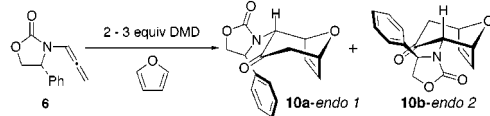
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(17) All new compounds were identified and characterized by <sup>1</sup>H NMR, <sup>13</sup>C NMR, FTIR, [α]<sub>D</sub><sup>20</sup>, and MS.

(18) An unidentifiable side-product was also isolated in 10–30% yield.

(19) Because DMD was generated in acetone, acetone was a cosolvent in all reactions reported here.

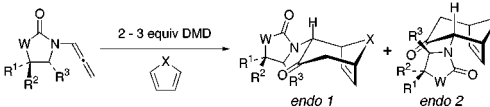
Table 1



entry	solvent <sup>a</sup>	temp, <sup>b</sup> °C	additive [2.0 equiv]	yield, <sup>c</sup> %	ratio of 10a:b <sup>d</sup>
1	CH <sub>2</sub> Cl <sub>2</sub>	-40	none	75	75:25
2	Et <sub>2</sub> O	-40	none	77	75:25
3	CH <sub>3</sub> CN	-40	none	<10	
4	THF	-40	none	80	75:25
5	THF	25	none	80	75:25
6	THF	-78	none	70	82:18
7	THF	-40	LiClO <sub>4</sub>	81	75:25
8	THF	-40	MgBr <sub>2</sub>	<10	
9	THF	25	ZnCl <sub>2</sub>	40	90:10
10	THF	-40	ZnCl <sub>2</sub>	77	94:6
11	THF	-78	ZnCl <sub>2</sub>	80	≥96:4

<sup>a</sup> Reaction solvent indicates the solvent that the allenamide **6** and 10.0 equiv of furan were dissolved in, although DMD was generated and added as a solution in acetone. <sup>b</sup> Reactions took 30 min at room temperature, 5–10 h at -45 °C, and 10–20 h at -78 °C to complete. <sup>c</sup> All yields are isolated yields. <sup>d</sup> Ratios were determined by using <sup>1</sup>H and/or <sup>13</sup>C NMR.

Table 2



entry <sup>a</sup>	allenes	W	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	X	adducts	yield, <sup>b</sup> %	endo ratio <sup>c</sup>
1	6	O	H	H	Ph	CH <sub>2</sub>	<b>11</b>	40	≥95:5
2	12	NMe	Me	H	Ph	O	<b>13</b>	60	≥95:5
3	12	NMe	Me	H	Ph	CH <sub>2</sub>	<b>14</b>	83	≥96:4
4	15	O	H	H	Bn	O	<b>16</b>	67	77:23
5	17	O	H	H	CHPh <sub>2</sub>	O	<b>18</b>	74	≥95:5
6	17	O	H	H	CHPh <sub>2</sub>	CH <sub>2</sub>	<b>19</b>	62	93:7
7	20	O	H	H	<i>i</i> -Pr	O	<b>21</b>	70	55:45
8	22	O	Ph	Ph	<i>i</i> -Pr	O	<b>23</b>	72	94:6

<sup>a</sup> Reactions were carried out in THF at -40 to -50 °C in the presence of 2.0–3.0 equiv of DMD [as a solution in acetone] and 10.0 equiv of the diene. For entries 1–3, 2.0 equiv of ZnCl<sub>2</sub> was used. All reactions were completed within 8 h. <sup>b</sup> All are isolated yields. <sup>c</sup> Endo ratios were determined by using <sup>1</sup>H and/or <sup>13</sup>C NMR.

The generality of this reaction is shown in Table 2. While reactions of chiral allenamides **6** or **12** with cyclopentadiene or furan were highly stereoselective [entries 1–3], **15** and **20**, containing benzyl and isopropyl groups, respectively, α to the nitrogen atom of the oxazolidinone ring, afforded much lower diastereoselectivities [entries 4 and 7]. Addition of ZnCl<sub>2</sub> did not improve diastereoselectivities but lowered reaction yields [not shown in Table 2]. However, chiral allenamide **17**, having a bulky dibenzylidene group α to the nitrogen atom [entries 5 and 6], improved the stereoselectivity from that of **15** even in the absence of ZnCl<sub>2</sub>, likewise with **22** that has the additional steric presence of the geminal diphenyl groups [entry 8].<sup>16b</sup> Attempts were made to explore the regioselectivity issue by using 2-methylfuran. However, reactions with 2-methylfuran were very slow leading to the cycloadduct in very low yield, and thus, it was not useful to determine the regioselectivity at this point.

A preliminary mechanistic working model was proposed in Figure 1 based on the stereochemical assignment. The chiral allenamide **6** [as an illustrative example] possesses two possible minimum conformations **A** and **B** based on AM1 calculations [Spartan Program]. The conformation **A** is favored owing to a

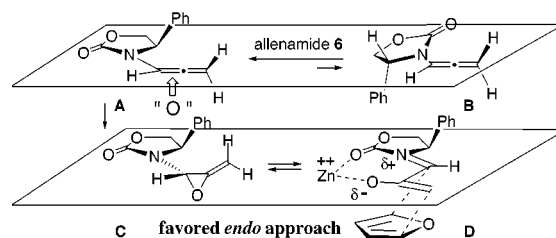
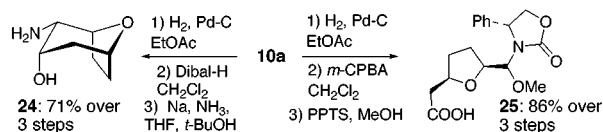


Figure 1.

Scheme 3



minimized dipole interaction, and the oxazolidinone ring is almost coplanar with the allenic moiety.<sup>16b</sup> Epoxidation could then occur at the bottom face away from the phenyl group, leading to the allene oxide **C** and subsequently to oxyallyl intermediate **D**.

The addition of dienes such as furan would likely proceed from the less congested bottom face of **D** away from the bulky phenyl ring, assuming **D** is involved in the cycloaddition.<sup>25</sup> The chelating ability of oxygen atoms to the Zn cation should enhance the conformational rigidity of the oxyallyl cation **D**, thereby leading to greater facial differentiation and diastereomeric induction. This model also lends support to the suggestion that nitrogen-substituted oxyallyl cations are essentially in a planar geometry.<sup>12</sup>

Finally, two examples are shown here in Scheme 3 to illustrate possible approaches for removal of the auxiliary,<sup>20</sup> thereby demonstrating the synthetic potential of these cycloadducts. First, hydrogenation and Dibal-H reduction followed by a Birch-type dissolving metal reduction<sup>21</sup> of the ketone **10a** led to the amino alcohol **24** [assigned by nOe experiments] stereoselectively in 71% yield over 3 steps.

Second, hydrogenation and Baeyer–Villiger oxidation<sup>22</sup> of **10a** followed by methanolysis led to the chiral aminal **25** as a single isomer [stereochemistry at the aminal carbocenter unassigned] in 86% overall yield. It is noteworthy that the Baeyer–Villiger oxidation of **10a** was both highly regio- and stereoselective.<sup>22</sup> Further hydrolysis of **25** via known methods<sup>23</sup> led to the corresponding aldehyde<sup>24</sup> and to recovery of the oxazolidinone auxiliary. Compounds **24** and **25** [possessing pseudosymmetry] represent excellent chiral building blocks for natural product synthesis.

We have described here the first epoxidations of 1-amidoallenes as a general entry to chiral nitrogen-substituted oxyallyl cation equivalents for stereoselective [4 + 3] cycloaddition. Efforts leading to synthetic applications of this methodology are currently underway.

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**Supporting Information Available:** Experimental procedures as well as <sup>1</sup>H/<sup>13</sup>C NMR spectra and characterization data for all new compounds (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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